

Chemical Composition and Antimicrobial Evaluation of *Blepharislinariifolia* Seeds**Abstract**

The current study has been designed to evaluate the bioactive compounds and antimicrobial activity of the n-hexane extract of *Blepharislinariifolia* seeds. The oils analyzed by GC-MS (Model GCMS-QP2010 Ultra, Shimadzu Co., Japan) which revealed the presence of 22 diversified compounds. All compounds were identified from NIST and WILEY spectral libraries. FTIR analysis revealed the presence of different functional groups. The in vitro antimicrobial assays showed there is no significant activity of this extract against five microorganism tested for.

Keywords: *Blepharislinariifolia*; oil; GCMS; FTIR; antimicrobial.

1. Introduction

Natural products are a well-known source of new valuable compound for medicinal or industry proposes [1]. Amongst the various natural sources, plants are the most important source of bioactive constituents including anticancer, antifungal and antimicrobial drugs. The use of plant compounds as prototypes of new drugs has a historical and economic importance[2-3].

The chemical analysis of plant extracts is extremely valuable to discover the chemical composition and to give better understanding of plant biological activities that may possess. Analysis of small amounts of chemicals has become easier and more accurate after the discover and development of chromatographic and spectroscopic techniques such as GC, HPLC, MS, NMR, etc. GC-MS are among the most powerful techniques used for both isolation and detection of samples which can be analyzed sufficiently even in trace amounts, less than 1 µg[4].

Fourier Transform Infrared Spectrophotometer (FTIR) is perhaps the most powerful tool for identifying the types of chemical bonds (functional groups) present in compounds. The wavelength of light absorbed is characteristic of the chemical bond as can be seen in the annotated spectrum. By interpreting the infrared absorption spectrum, the chemical bonds in a molecule can be determined[5].

31 *B. linariifolia*(family: Acanthaceae) distributed in Africa especially the areas from Mauritania to
32 Sudan, through Arabia to Northwestern India.*B. linariifolia* is a low-growing, wiry herb with
33 prickly bracts and blue flowers. Seeds have analgesic activity, are also used in veterinary
34 medicine [6]. however, genus *Blepharis* have not investigated much[7].

35 **2. Materials and methods**

36 **2.1. Extraction**

37 The fresh samples were dried in shades for 7 days, powdered then used for extraction. Cold
38 maceration methodology was used and it was carried out according to published method of
39 Osama and Awdelkarim, 2015 [8].

40 **2.2. Fourier Transform Infrared Spectrophotometer (FTIR) analysis**

41 Dried powder sample was used for FTIR analysis using KBr disk methodology. 1 mg of
42 sample was encapsulated in 100 mg of KBr pellet in order to prepare translucent sample discs.
43 The powdered sample was loaded in FTIR spectroscope (Shimadzu, IR Affinity 1, Japan), with a
44 scan range from 400 to 4000 cm^{-1} with a resolution of 4 cm^{-1} . Each analysis was repeated ten
45 times for the spectrum confirmation

46 **2.3. GC-MS Analysis**

47 GC-MS analysis was carried out by using the GC instrument (Model GCMS-QP2010 Ultra,
48 Shimadzu Co., Japan) equipped with a capillary column Rtx-5 (0.25 μm film \times 0.25 mm i.d. \times 30
49 m length). The instrument was operated in electron impact mode at ionization voltage (70 eV),
50 injector temperature (250 $^{\circ}\text{C}$), and detector temperature (280 $^{\circ}\text{C}$). The carrier gas used was
51 helium (99.9% purity) at a flow rate of 1.2 mL/min and about 1 μL of the sample was injected.
52 The oven temperature was initially programmed at 35 $^{\circ}\text{C}$ (3 min) to 240 $^{\circ}\text{C}$ at 5 $^{\circ}\text{C}/\text{min}$ and from
53 240-280 $^{\circ}\text{C}$ at 3 $^{\circ}\text{C}/\text{min}$ withhold time and 4 min respectively. The identification of
54 compounds from the spectral data was based on the available mass spectral records (NIST and
55 WILEY libraries).

56 **2.4. Antimicrobial evaluation**

57 **2.4.1. Preparation of bacterial suspensions**

58 One ml aliquots of a 24 hours broth culture of the test organism were aseptically distributed
59 onto nutrient agar slopes and incubated at 37 $^{\circ}$ C for 24 hours. The bacterial growth was harvested
60 and washed off with 100 ml sterile normal saline, to produce a suspension containing about 108-
61 109 c.f.u/ml. The suspension was stored in a refrigerator at 4 $^{\circ}$ C till used. The average number of

62 viable organisms per ml of the stock suspension was determined by means of the surface viable
63 counting technique. Serial dilutions of the stock suspension were made in sterile normal saline
64 solution and 0.02 ml volumes on drop of the appropriate dilutions were transferred by micro
65 pipette onto the surface of dried nutrient agar plates. The plates were allowed to stand for two
66 hours at room temperature for the drops to dry and then incubated at 37 °C for 24 hours. After
67 incubation, the number of developed colonies in each drop was counted. The average number of
68 colonies per drop (0.02 ml) was multiplied by 50 and by the dilution factor to give the viable
69 count of the stock suspension, expressed as the number of colony forming units per ml
70 suspension. Each time a fresh stock suspension was prepared. All the above experimental
71 conditions were maintained constant so that suspensions with very close viable counts would be
72 obtained.

73 **2.4.2. Preparation of fungal suspension**

74 The fungal culture were maintained on Sabouraud dextrose agar, incubated at 25 °C for 4 days.
75 The fungal growth was harvested and washed with sterile normal saline and finally suspension in
76 100 ml of sterile normal saline, and the suspension were stored in the refrigerator until used.

77 **2.4.3. In vitro testing of extracts for antimicrobial activity**

78 **A) Testing for Antibacterial Activity**

79 The cup-plate agar diffusion method was adopted according to published method of Eltayeb et
80 al., 2015 [9] with some minor modification assess the antibacterial activity of the prepared
81 extract. One ml of the standardized bacterial stock suspension 10^8 – 10^9 C.F.U./ml were
82 thoroughly mixed with 100ml of sterile molten nutrient agar which was maintained at 45 °C.
83 20ml aliquots of the inoculated nutrient agar were distributed into sterile Petri-dishes. The agars
84 was left to dry and in each of these plates 4 cups (10 mm in diameter) were cut using a sterile
85 cork borer (No. 4) and agar discs were removed. Alternate cups were filled with 0.1 ml of
86 each extracts using automatic Microlitre-pipette, and allowed to diffuse at room temperature for
87 two hours. The plates were then incubated in the upright position at 37 °C for 18 hours. Two
88 replicates were carried out for each extracts against each of the test organisms. Simultaneously
89 addition of extracts was carried out as controls. After incubation, the diameters of the resultants
90 and growth inhibition zones were measured, averaged and the mean values were tabulated.

91 **b) Testing for Antifungal Activity**

92 The same method as for bacteria was adopted. Instead of nutrient agar, Sabouraud dextrose agar
 93 was used. The inoculated medium was incubated at 25 °C for three days for *Aspergillus niger*.

94 **3. Results and discussion**

95 The n-hexane was able to dissolve about 2.7 % of total sample mass. Hexane is a nonpolar
 96 solvent capable to isolate the nonpolar compounds especially fatty acids, terpenes and steroids.
 97 This low yield percentage indicates the low amount of oil in this plant. The nature of sample is
 98 not the only factor that affect its content of group of compounds. The environmental factors, the
 99 part of plant used, methodology of extraction, duration of extraction, solvent used for extraction,
 100 etc. all these are factors can effect on percentage yield [10].

101 In the current study the oil was analyzed using spectrometric techniques (FTIR and GCMS) to
 102 determine the functional groups and chemical constituents. The FTIR is the most powerful
 103 technique to determine the functional groups according to the response of compounds to the
 104 radiation with adjusted wave length. These groups are most likely responsible for the chemical
 105 and biological activities of this extract. The functional groups are the active parts of compounds
 106 they affect to its reactivity to certain kind of compounds or receptors. And the study of the
 107 structure and its activities is the base of valuable part of medicinal chemistry which named as
 108 structure activity relationship [11]. The results showed the presence of different groups, shown in
 109 table (1).

110

111 **Table 1. Functional groups and its wave number.**

cm ⁻¹	Bond	functional group
721.33	C–X stretching (X = F, Cl, Br or I)	organic halogen
1149.50	Aliphatic C–O stretching	Ester
1377.08	Aliphatic NO ₂ symmetric stretching	nitro compound
1460.01	C–C stretching	(in–ring) aromatics
1537.16	N–O asymmetric stretching	nitro compounds
1581.52	N–H bend	1° amines
1668.31	C=O stretching	carbonyls (general)
1822.61	Overtone and combination bands	Other
2352.99	Combination C–H stretching	Common near-infrared bands of organic compounds

2731.02	C–H stretching	Aldahyde
2866.02	symmetric C–H stretching	Alkenes
2925.81	asymmetric C–H stretching	Alkenes
2958.60	symmetric C–H stretching	Alkenes
3544.92	O–H stretching	Phenol
3589.28	O–H stretching	Alcohol
3627.85	O–H stretch	free hydroxyl alcohols, phenols
3652.93	O–H stretch,	free hydroxyl alcohols, phenols
3670.28	O–H stretching	free hydroxyl alcohols, phenols
3712.72	O–H stretching	free hydroxyl alcohols, phenols
3780.22	Si–OH stretching	silicon compounds
3876.65	O–H stretching	free hydroxyl alcohols, phenols
3926.80	O–H stretching	free hydroxyl alcohols, phenols

112

113 The functional groups detected by FTIR was mainly belong to aliphatic hydrocarbons, the
 114 highest absorption was found in 2958.6 (symmetric C–H stretching) which indicates the high
 115 amount of hydrocarbons alkenes. The hydrocarbons are well known for their non-polar
 116 properties therefore they can be isolated by hexane, the solvent used for extraction. However, the
 117 alkenes are not active compounds compared to phenols or carboxylic acids. Low absorption was
 118 noticed for OH and C=O groups which are known for their antimicrobial activity.

119 Gas chromatography- mass spectrometry is one of the most updated techniques to isolate and
 120 detected the volatile chemical substances. In the present study the GCMS analysis detect
 121 different kind of compounds with variable molecular weight. The chromatogram showed the
 122 presence of 22 compounds, which they elucidated by MS. The MS showed that the molecular
 123 weight.of these compounds which vary from 100 to 506 amu. These compounds are mainly
 124 aliphatic hydrocarbons, shown in table (2).

125 **Table 2. GCMS analysis of *B. linariifolia***

Peak no.	R. Time	Area %	Compounded name	Molecular Formula	Mass
1	4.655	0.44	2-Pentanol, 2-methyl	C6H14O	102
2	5.157	0.22	3-Pentanol, 3-methyl	C6 H14 O	102
3	6.199	0.46	2-Hexanone	:C6H12O	100
4	6.474	0.72	Hexanal	C6H12O	100

5	11.685	0.57	Benzaldehyde, 4-fluoro	C7H5FO	124
6	19.309	0.50	4-Methyl-4-(tetrahydropyran-2-yl)oxypentane-2,3-dione	C11H18O4	214
7	19.643	0.16	4-methyl-4-[3',4',5',6'-tetrahydro-2'-H-pyran-2'-oxy]-2,3-pentanedione	C11H18O4	214
8	19.988	0.34	Hexane, 1,1'-oxybis	C12H26O	186
9	20.064	0.19	1-Pentanol, 2,2-dimethyl-	C7H16O	116
10	20.273	0.52	Butanoic acid, 2-ethyl-2-methyl	C7H14O2	130
11	21.346	2.18	2-Heptene, 5-ethyl-2,4-dimethyl	C11H22	154
12	23.226	1.94	Undeca-4,8-dione	C11H20O2	184
13	29.413	11.60	Acetic acid, 4-acetyl-2-isopropyl-5,5-dimethyltetrahydrofuran-2-yl ester	C13H22O4	242
14	31.695	3.10	1-(3,3-dimethyl-bicyclo[2.2.1]hept-2-yl)pentan-2-one	C14H24O	208
15	32.078	13.09	3-cyano-2-oxa-1-ethoxyadamanane	C12H17NO2	207
16	34.041	15.49	Ethyl 3-methyl-2-oxobutyrate	C7H12O3	144
17	43.429	1.56	Heneicosane	C21H44	296
18	44.396	0.58	4,8,12,16-Tetramethylheptadecan-4-olide	C21H40O2	324
19	46.706	2.13	1,2-Benzenedicarboxylic acid, diisooctyl ester	C24H38O4	390
20	47.756	2.87	Hexatriacontane	C36H74	506
21	50.337	8.18	Hexatriacontane	C36H74	506
22	54.058	16.03	Dotriacontane	C32H66	450

126

127 The *B. linariifolia* oil was evaluated for its antimicrobial activity and was found to be
 128 inactive to five type of bacteria and fungal. Table (3). Compounds or plant extracts are
 129 considered to be active if the inhibition zone exceeded 15 mm[12].

130 **Table 3. Antimicrobial evaluation of *B. linariifolia* against five standard organisms.**

Type of microbe	Gram +ve		Gram -ve		Fungal
	<i>Bacillus subtilis</i>	<i>Staphylococcus aureus</i>	Escherichia coli	<i>Pseudomonas aeruginosa</i>	<i>Aspergillusniger</i>
MDIZ (mm)	11	11.5	12	14	zero

131 MDIZ = mean diameter of growth inhibition zone

132 **4. Conclusion** 

133 In conclusion the *B. linariifolia* oil was analyzed with FTIR and GCMS also was evaluated for
134 their antimicrobial activity against four type of bacteria and a fungal. Different functional groups
135 were detected. The main type of compounds was found to be alkenes. The GCMS revealed the
136 presence of variable compounds with diverse structures. Low antimicrobial activity was noticed.
137 Further studies for both chemical and biological proposes are recommended.

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